

On the Relative Importance of SO₂ oxidation to High Dust SCR DeNOx units

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Outline

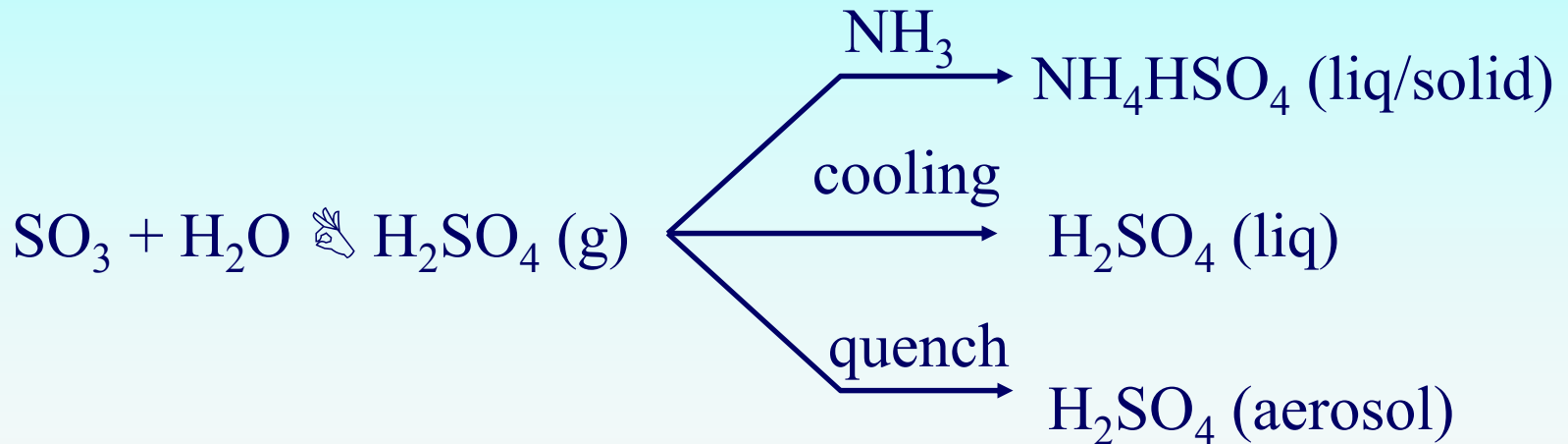
- **Introduction**
 - SO_3 formation and related problems
- **SO_3 capture by fly ash**
 - Prediction of SO_3 concentration at SCR outlet
 - Model description and results
- **SO_3 sampling technique**
 - A critical review of the Controlled Condensation Method
 - SO_3 capture by fly ash on filter
 - SO_3 collection
 - SO_4^{2-} determination
- **Case studies**
 - The fate of SO_3 in coal fired power plants
 - Sampling related problems
 - Results from two U.S. coal fired power plants
 - Harrison Power Plant (3 x 640 MW)
 - Pleasants Power Plant (2 x 625 MW)
- **Conclusions**



SO₃ formation and related problems

SO₃ formation

- Furnace 0.7 % of SO₂ conc.
 - rule of thumb (US-EPA)
- SCR - SO₂ oxidation 0.5-1 % of SO₂ conc.
- Total 1.2-1.7 % of SO₂ conc.



SO₃ capture by fly ash

Less SO₂ oxidation in power plant SCR DeNOx compared to dust free laboratory measurements



Prediction advantages:

- More accurate SO₃ concentrations at the SCR outlet
- Air heater optimization
- Effect of fuel switching or SCR retrofitting



SO₃ capture by fly ash - model

The SO₃ removal rate in the flue gas follows a simple reaction mechanism:

$$-r_{\text{SO}_3} = k \cdot C_{\text{SO}_3} \cdot C_{\text{CaO}}$$

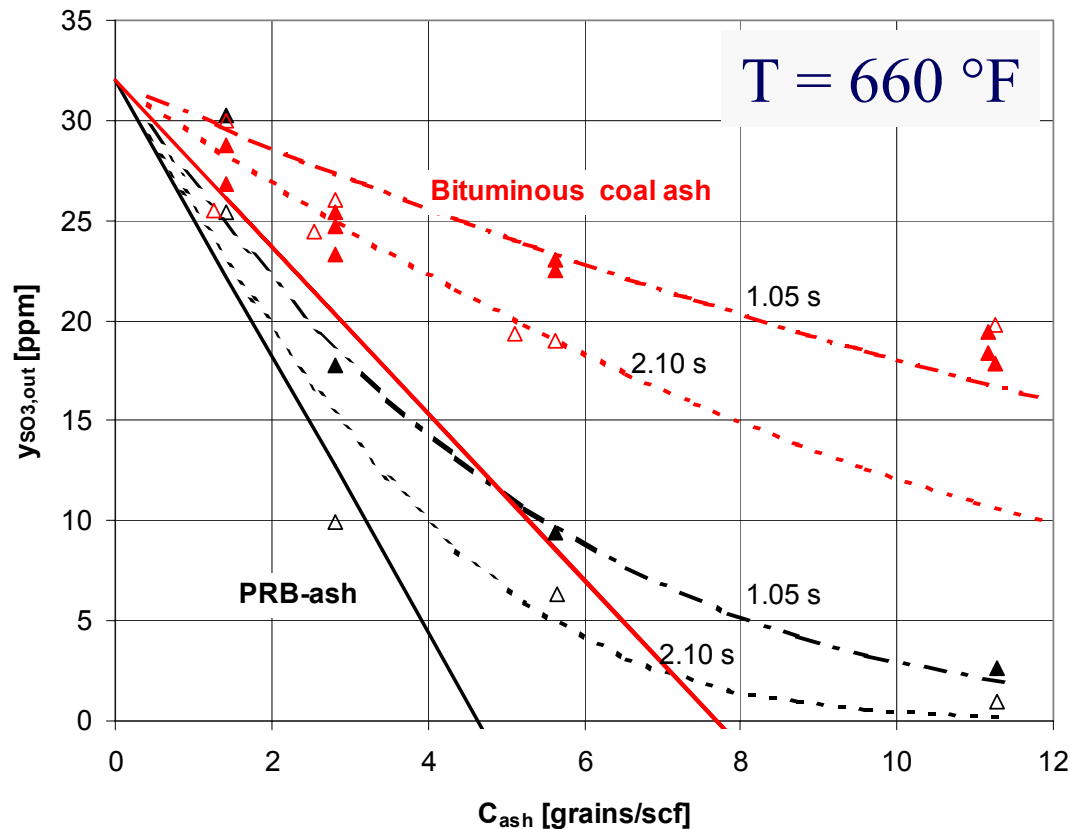
Reaction constant
→ temperature

Gaseous SO₃
concentration

Concentration of alkaline sites
→ ash composition
→ ash concentration
→ particle size distribution



SO₃ capture by fly ash - model



Pilot plant experients

- dust type
- dust loading
- residence time
- temperature
- SO₃ conc.

PRB-ash

- Alkaline
- Fines

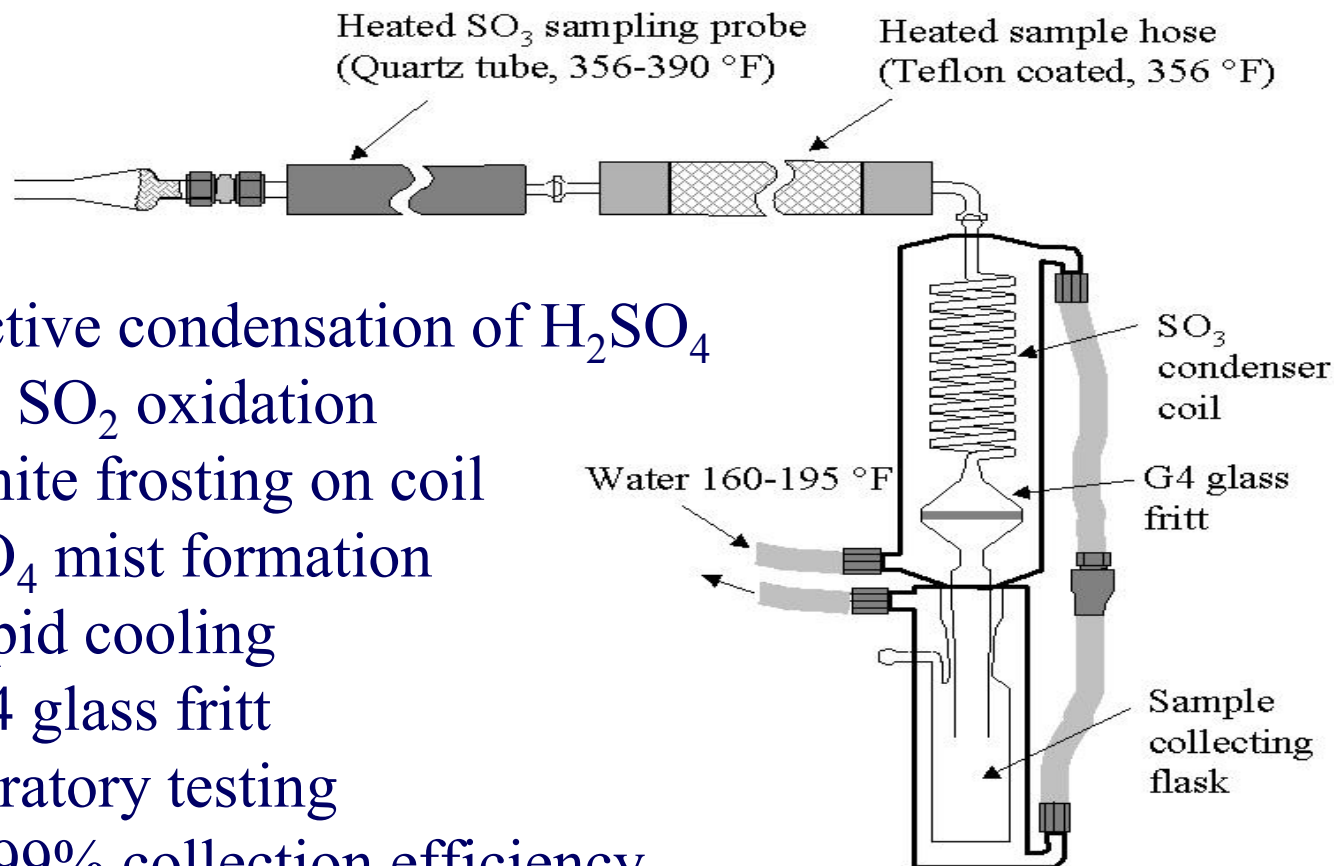


SO₃ measurements

Controlled Condensation Method- a critical review



SO₃ measurements - collection



Selective condensation of H₂SO₄

- no SO₂ oxidation
- white frosting on coil

H₂SO₄ mist formation

- rapid cooling
- G4 glass frit

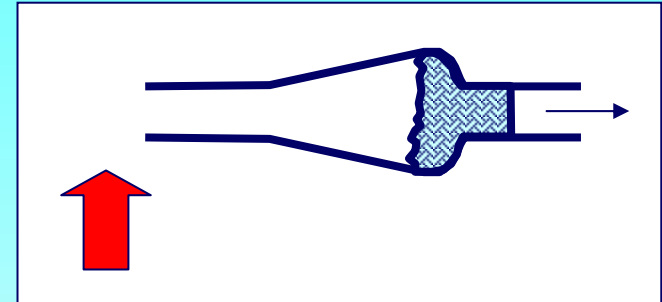
Laboratory testing

- > 99% collection efficiency

SO₃ measurements - ash filtration

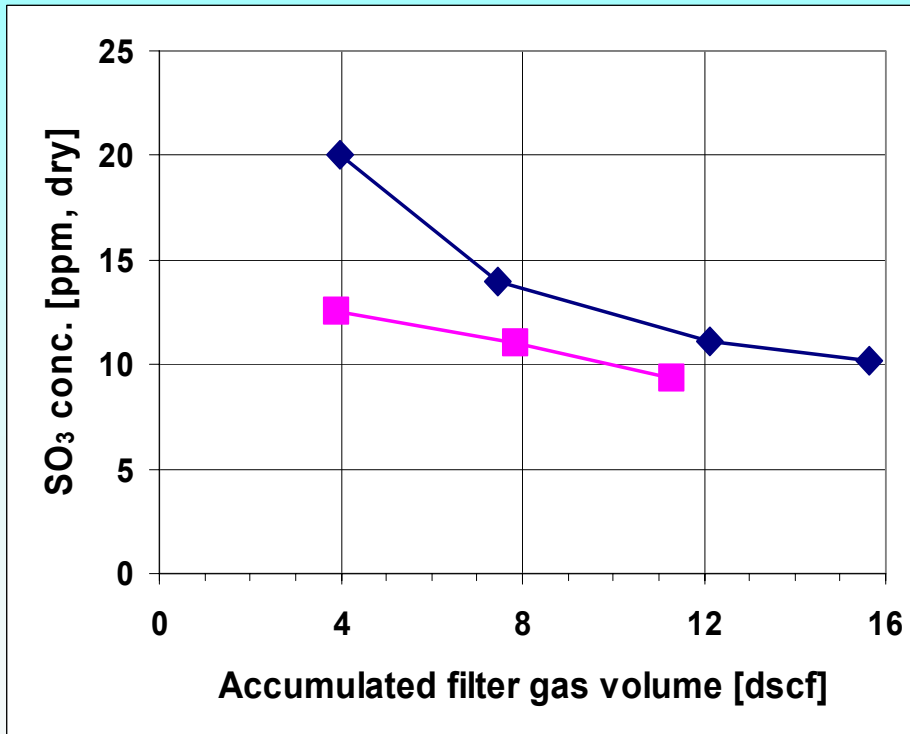
Coal fired power plant

- bituminous coal
- outlet of two SCR's
- 750 °F flue gas



Sampling

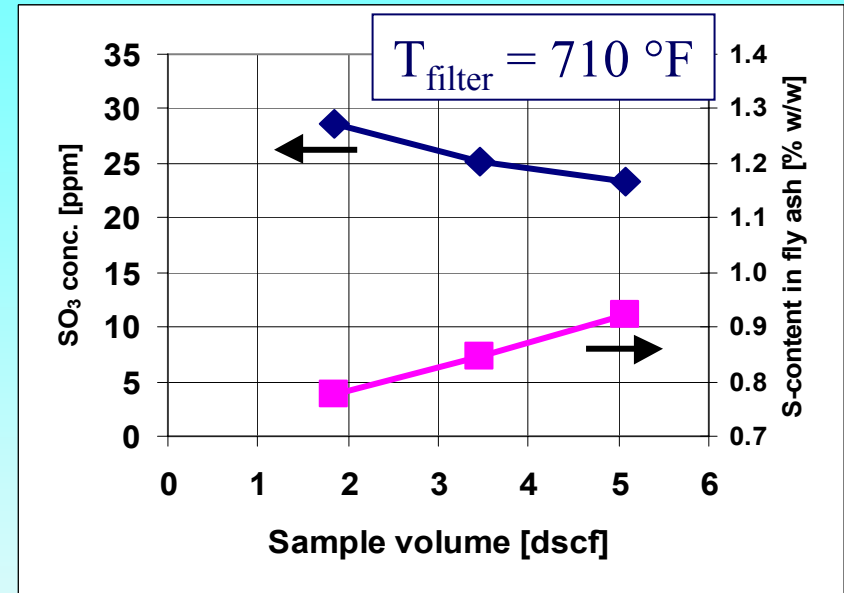
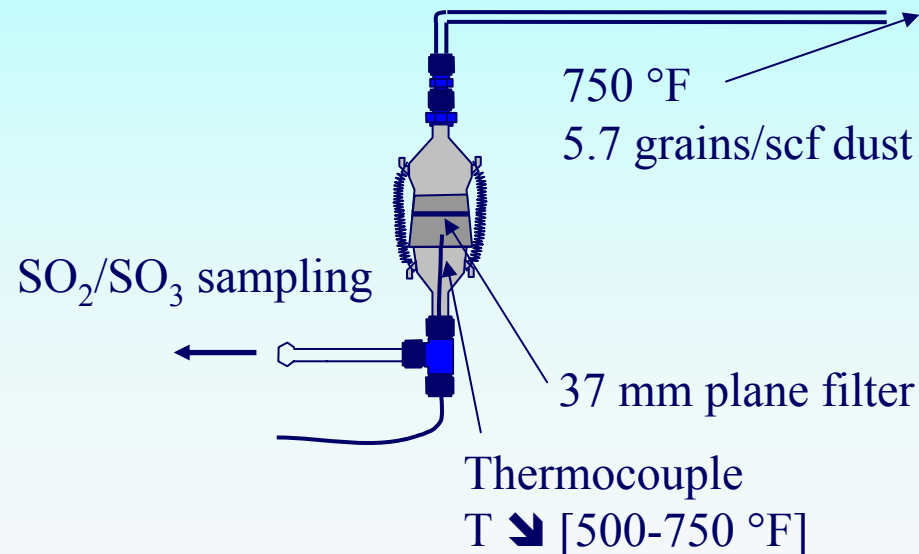
- in-duct quartz wool plug
- perpendicular sampling
- re-using filter in consecutive measurements



SO₃ measurements - ash filtration

Pilot plant experiments

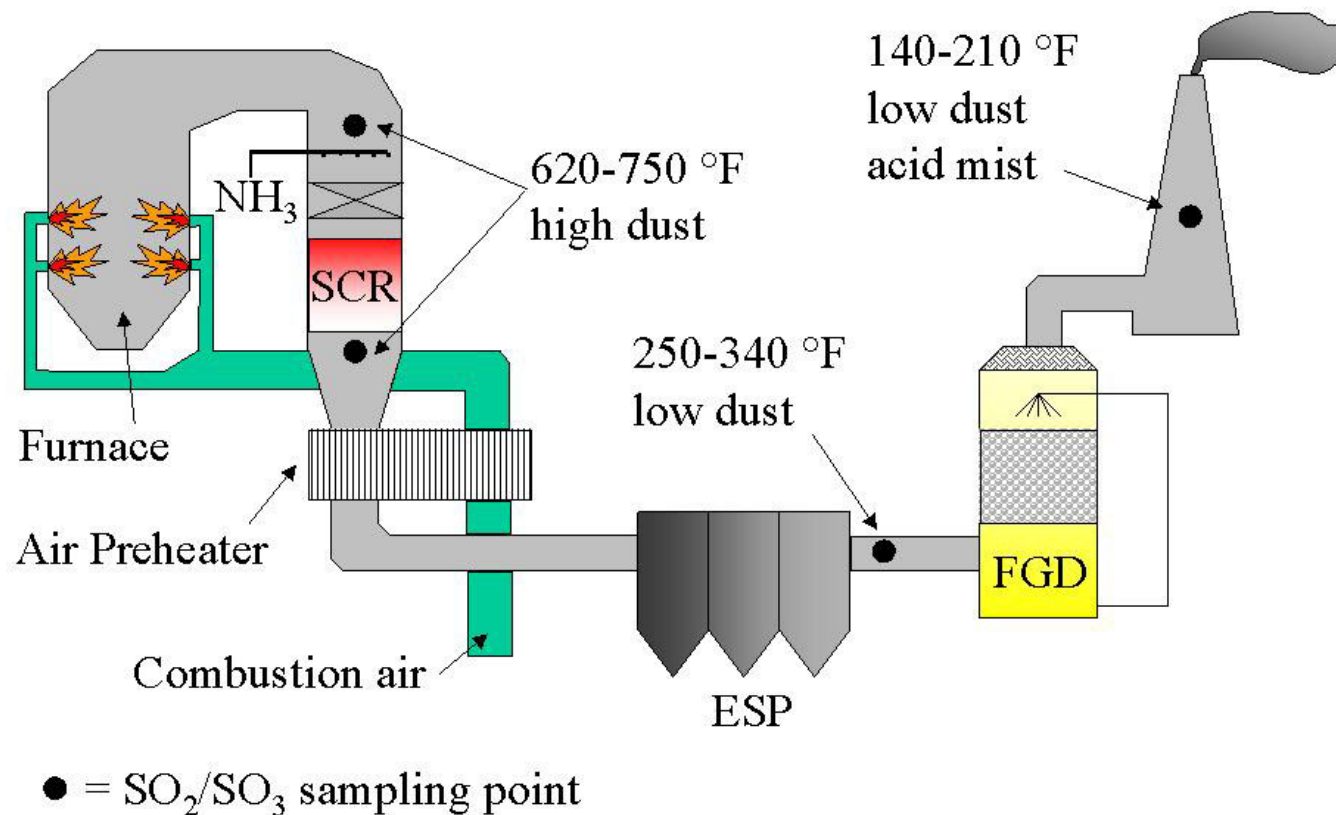
- bituminous coal ash
- 32 ppm SO₃ inlet
- quartz plane filter
- filter load 2 grains/ft³



6 ppm SO₃ (~20%)
capture per 3.5 dscf
sample volume,
independent on T_{filter}



Case studies - the fate of SO_3



Case studies - power plant data

Harrison power station

- 3 x 640 MW
- high sulfur bituminous coal
- SCR deNOx
- rotary air heater
- cold side ESP
- wet SO₂ scrubber
 - sulfite precipitation

Pleasants power station

- 2 x 625 MW
- high sulfur bituminous coal
- SCR deNOx
- rotary air heater
- cold side ESP
- wet SO₂ scrubber
 - forced oxidation



Case stories - results

SCR DeNOx

- ☞ Double determinations in three sampling ports
- ☞ Four SCR units tested on each power plant
- ☞ >12 hour stabilization time
- ☞ Static mixers upstream sample ports at Harrison

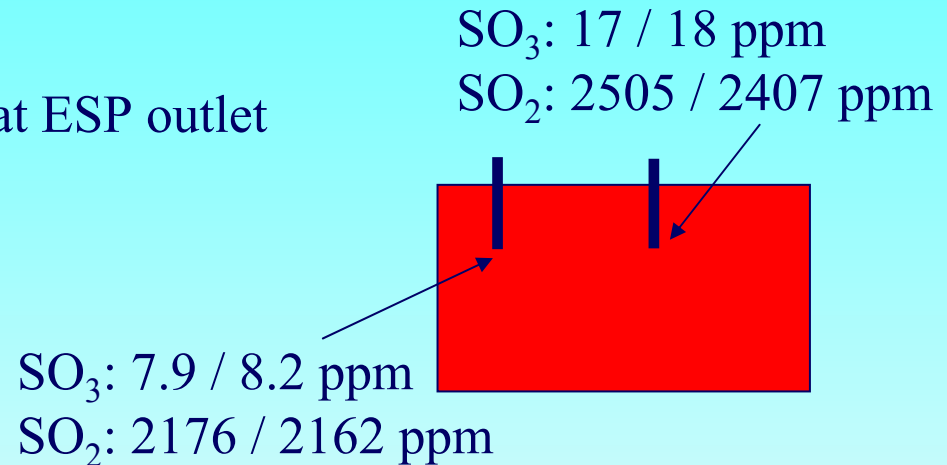
	Harrison	Pleasants
— SO ₂ in	2323 - 2692 ppm	2740 - 3254 ppm
— SO ₃ in	26 - 30 ppm	19 - 37 ppm
— inlet stratification		
• SO ₂	2.8 - 5.2 %	1.4 - 7.3 %
• SO ₃	7 - 15 %	5 - 42 %
— SO ₂ out / SO ₂ in	0.98 - 1.09	0.96 - 1.01
— SO ₂ oxidation	0.29 - 0.81 %	0.42 - 0.92 %
• Temperature	630 - 678 °F	658 - 696 °F



Case stories - results

ESP and Stack

- Significant SO₂/SO₃ stratification at ESP outlet
 - ➔ rotary air heater operation
- Some SO₃ removal in FGD



	Harrison	Pleasants
— SO ₂ ESP outlet	2313 ppm	2370 ppm
— SO ₃ ESP outlet	13 ppm	40 ppm
— stratification		
• SO ₂	7.4 %	14 %
• SO ₃	46 %	7.5 %
— SO ₃ in stack	10 ppm	22 ppm



Conclusions

- **SO₃ capture by fly ash reduces observed SO₂ oxidation across SCR DeNOx**
 - Prediction by simple reaction mechanism
- **SO₃ sampling is not simple**
 - SO₃ capture by fly ash on filter
 - Risk of insufficient SO₃ collection
 - Underestimation of true SO₃ concentration
- **SO₃ emissions depends on the overall operation of the power plant**
 - furnace, SCR deNOx, rotary airheater, ESP and FGD

